Mahila Vikas Sanstha's



New Arts, Commerce & Science College, Wardha

(Accredited 'B' by NAAC)

* Approved by Government of Maharashtra.

- * Affiliated to Rashtrasanta Tukdoji Maharaj Nagpur University, Nagpur.
- * Recognised by U.G.C. New Delhi Under section 2(f) & 12 (b) of UGC act 1956.

Batchelor Road, Nalwadi Wardha - 442 001. (Maharashtra) 2 : (07152) - 253093,240152, Fax : 07152-240152

Class : BSc 1 semester - 2019

Subject : Inorganic Chemistry

Paper 1

Unit 1

- 1. Explain Hund's rule of maximum multiplicity
- 2. State Aufbau principle and explain its importance
- 3. Define ionization potential
- 4. Discuss factors affecting ionization potential and how it varies in group and period
- 5. Write Schrodinger wave equation and specify the terms involved in it
- 6. Give the values of n, l, m for 3d electron
- 7. Why the size of cation and anion of an element is different from its parent atom
- 8. Calculate the effective nuclear charge for 4s electron of Potassium
- 9. Define electron affinity and factors affecting it
- 10. Draw shapes of dx2-y2

Unit 2

- 1. Define hybridization. Discuss sp3 and sp3d2 hybridization with suitable example
- 2. Define polarizing power of ion
- 3. Define polarizability of ion
- 4. Discuss various factors affecting polarizing power and polarizability
- 5. What are the postulates of VBT
- 6. define the term bond energy
- 7. What is solvation energy? How solvation energy of NaCl is determined using Born-Haber cycle
- 8. Discuss structure of CsCl
- 9. Mention two limitations of VBT
- 10. Draw structure of SF6

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Unit 3

- 1. Write electronic configuration of s block elements
- 2. Compare ionization energy of s block elements
- 3. Discuss structure and bonding in XeF2
- 4. Give one method of preparation of XeF2 and XeF4
- 5. Define hydrogen bond. How does it affect melting and boiling points of compounds
- 6. What is diagonal relationship
- 7. Discuss diagonal relationship between Lithium and Magnesium
- 8. Discuss structure and bonding in XeF6
- 9. Define lattice energy
- 10. Give one example of intramolecular hydrogen bonding

Unit 4

- 1. Discuss the periodicity of p block elements
- 2. Discuss the conept of electronegativity for p block elements
- 3. What are boranes? Discuss the structure and bonding in diborane in detail
- 4. Discuss the structure of P2O5
- 5. Give method of preparation of Marshalls acid
- 6. Explain the structure of Caro's acid
- 7. Discuss diagonal relationship between Boron and silicon
- 8. Discuss the structure of orthophosphoric acid
- 9. Why first IP of nitrogen is higher than oxygen
- 10. What is borazine